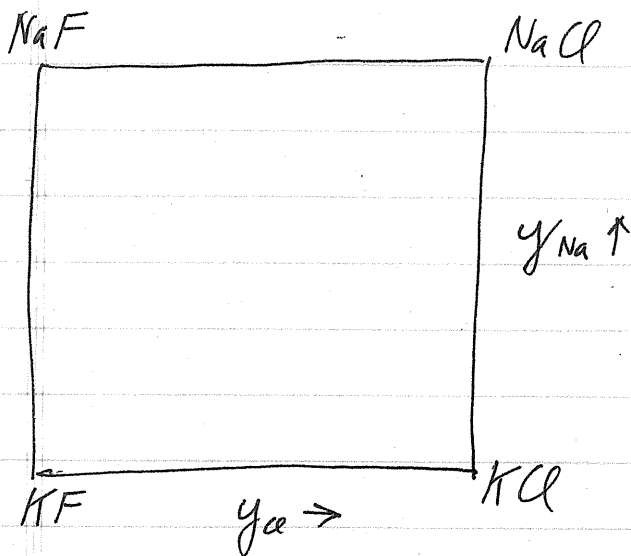


EQUATIONS (I-119) to (I-122)
in more recent notation



$$y_i = \text{site fraction}$$

$$y_{Na} = \frac{n_{Na}}{n_{Na} + n_K} = (1 - y_K)$$

$$y_{Cl} = \frac{n_{Cl}}{n_{Cl} + n_F} = (1 - y_F)$$

$$\Delta G^{\text{exchange}} = (g_{NaF}^0 + g_{KCl}^0 - g_{NaCl}^0 - g_{KF}^0)$$

$$g = (y_{Na} y_{Cl} g_{NaCl}^0 + y_K y_F g_{KF}^0 + y_{Na} y_F g_{NaF}^0 + y_K y_{Cl} g_{KCl}^0)$$

$$\left[\begin{aligned} &+ RT (y_{Na} \ln y_{Na} + y_K \ln y_K) \\ &+ RT (y_F \ln y_F + y_{Cl} \ln y_{Cl}) \end{aligned} \right.$$

$$\left[\begin{aligned} &+ y_{Na} (y_F y_{Cl} \alpha_{NaF-NaCl}) + y_K (y_F y_{Cl} \alpha_{KF-KCl}) \\ &+ y_F (y_{Na} y_K \alpha_{NaF-KF}) + y_{Cl} (y_{Na} y_K \alpha_{NaCl-KCl}) \end{aligned} \right.$$

$$- y_{Na} y_K y_F y_{Cl}$$

where:

$$\alpha_{NaF-NaCl} = \sum_{i=0}^i L_{Na, F, Cl} (y_F - y_{Cl})^i$$

$$\alpha_{KF-KCl} = \sum_{i=0}^i L_{K, F, Cl} (y_F - y_{Cl})^i$$

$$\alpha_{NaF-KF} = \sum_{i=0}^i L_{Na, K, F} (y_{Na} - y_K)^i$$

$$\alpha_{NaCl-KCl} = \sum_{i=0}^i L_{Na, K, Cl} (y_{Na} - y_K)^i$$

$$L = (\Delta G^{\text{exchange}})^2 / ZRT$$